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of water occurs
when two water
molecules
produce a
hydronium (H_3O^+)
and a hydroxide
(OH^-) ion $2 \text{H}_2\text{O}$
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of water occurs

when two water

molecules

produce a

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and a hydroxide

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(OH⁻) ion 2 H₂O

H₃O⁺ + OH⁻ The
ionization

constant (K_w) of

water is: K_w =

[H₃O⁺] [OH⁻] =

1.0 x 10⁻¹⁴ M

Acidic, Basic, &

Neutral IF

[H₃O⁺] >

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REVIEW Acid-Base
Titration and pH

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the following
questions in the
space provided.

1. Below is a pH
curve from an
acid-base
titration. On it
are labeled
three points: X,

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Chemistry, and Z. a.

Which point
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represents the
equivalence

point? b. At
which point is
there excess
acid in the
system? c.

~~15 Acid Base~~

~~Titration and pH~~

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set (10) Self-

ionization of

water. two water

molecules

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produce a hydronium ion and a hydroxide ion by transfer of a proton. pH . the negative of the common logarithm of the hydronium ion concentration. pOH .

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answer to "Which
of the following
masses is the
largest?"

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Solutions 1

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Section 1- Types

of Mixtures.

Solutions are

homogeneous

mixtures of two

or more

substances in a

single phase.

Soluble

describes a

substance as

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capable of being dissolved.

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