

## Section 8 Covalent Bonding Answers

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240 Chapter 8 • Covalent Bonding Section 88.1.1 The Covalent Bond-1.)DEAAAtoms gain stability when they share electrons and form covalent bonds. Real-World Reading Link Have you ever run in a three-legged race? Each person in the race shares one of their legs with a teammate to form a single three-legged team.

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Chapter 8 Covalent Bonding and Molecular Structure 8-3 There are two types of repulsive forces between the two atoms. First, the nuclei repel because they are both positively charged. Second, the electrons repel because they are both negatively charged. The attractive forces between the two atoms result from the

Chapter 8: Covalent Bonding and Molecular Structure  
Section 8.4 – Polar Bonds and Molecules. Covalent bonds involve sharing electrons between atoms. When the atoms in the bond pull equally, the bonding electrons are shared equally, and the bond is nonpolar. When the atoms in the bond pull unequally, the bonding electrons are pulled closer to one atom, and the bond is polar.

Chapter 8 – Covalent Bonding  
Covalent bond A bond formed by the sharing of electrons between atoms.

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Download Ebook Chapter 6 Covalent Bonding Answers Chapter 6.2 : Covalent Bonding and Molecular Compounds 6. Define a polar covalent bond. 7. When atoms form a polar covalent bond, the atom with the greater attraction for electrons has a partial charge.

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SECTION 8.1 The Covalent Bond Single Covalent Bonds (cont.) • Atoms in group 16 can share two electrons and form two covalent bonds. • Water is formed from one oxygen with two hydrogen atoms covalently bonded to it.

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